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APPLICATION FOR UNITED STATES LETTERS PATENT

for

OIL BASED WELL FLUIDS WITH HIGH SOLIDS CONTENT

by

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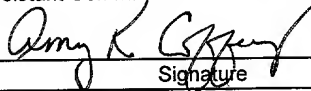
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BACKGROUND OF THE INVENTION

Many types of fluids have been used in well bores particularly in connection with the drilling of oil and gas wells. The selection of an oil-based well bore fluid involves a careful balance of the both the good and bad characteristics of such fluids in a particular application. The primary benefits of selecting an oil-based drilling fluid include: superior hole stability, especially in shale formations; formation of a thinner filter cake than the filter cake achieved with a water based mud; excellent lubrication of the drilling string and downhole tools; penetration of salt beds without sloughing or enlargement of the hole as well as other benefits that should be known to one of skill in the art. An especially beneficial property of oil-based muds is their excellent lubrication qualities. These lubrication properties permit the drilling of wells having a significant vertical deviation, as is typical of off-shore or deep water drilling operations or when a horizontal well is desired. In such highly deviated holes, torque and drag on the drill string are a significant problem because the drill pipe lies against the low side of the hole, and the risk of pipe sticking is high when water based muds are used. In contrast oil-based muds provide a thin, slick filter cake which helps to prevent pipe sticking and thus the use of the oil-based mud can be justified.

Despite the many benefits of utilizing oil-based muds, they have disadvantages. In general, the use of oil based drilling fluids and muds has high initial and operational costs. These costs can be significant depending on the depth of the hole to be drilled. However, often the higher costs can be justified if the oil based drilling fluid prevents the caving in or hole enlargement which can greatly increase drilling time and costs.

In general, drilling fluids should be pumpable under pressure down through strings of drilling pipe, then through and around the drilling bit head deep in the earth, and then returned back to the earth surface through an annulus between the outside of the drill stem and the hole wall or casing. Beyond providing drilling lubrication and efficiency, and retarding wear, drilling fluids should suspend and transport solid particles to the surface for screening out and disposal. In addition, the fluids should be capable of suspending additive weighting agents (to increase specific gravity of the mud), generally

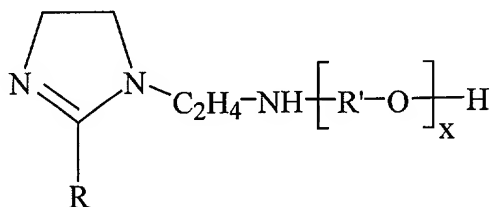
1 finely ground barites (barium sulfate ore), and transport clay and other substances capable
2 of adhering to and coating the borehold surface.

3 Drilling fluids are generally characterized as thixotropic fluid systems. That is
4 they exhibit low viscosity when sheared, such as when in circulation (as occurs during
5 pumping or contact with the moving drilling bit. However, when the shearing action is
6 halted, the fluid should be capable of suspending the solids it contains to prevent gravity
7 separation. In addition, when the drilling fluid is under shear conditions and a free-
8 flowing near-liquid, it must retain a sufficiently high enough viscosity to carry all
9 unwanted particulate matter from the bottom of the well bore to the surface. The drilling
10 fluid formulation should also allow the cuttings and other unwanted particulate material
11 to be removed or otherwise settle out from the liquid fraction.

12 Thus a search has been going on for many years for an improved additive for
13 modifying and controlling the solids tolerance properties of drilling fluids that will
14 provide fluids which are efficient, easily handled, and readily dispersible, and usable
15 under high solids contamination and other broad range of conditions.

17 SUMMARY OF THE INVENTION

18 The present invention is generally directed to a fluid having utility in subterranean
19 wells having improved solids tolerance capabilities. One such illustrative fluid includes
20 an oleaginous fluid; and a solids tolerance agent having the formula:



21
22 in which R is a C₆ to C₂₀ aliphatic group and R' is a C₂ to C₆ aliphatic group and x
23 has a value from about 1 to about 10. In one preferred embodiment of the above fluid,
24 the R' group is ethyl or isopropyl. In another preferred embodiment, the R is unsaturated
25 having at least one carbon-carbon double bond in it's carbon chain. Preferably, the
26 oleaginous fluid is from about 30% to about 99% by volume of the total of the fluid. In

1 one illustrative and preferred embodiment, the oleaginous fluid is composed of from
 2 about about 5% to about 100% by volume of the oleaginous fluid of a material selected
 3 from diesel oil, mineral oil, synthetic oil, esters, ethers, acetals, di-alkylcarbonates,
 4 olefins, combinations of these and other similar fluids used in the formulation of well
 5 bore fluids.

6 Optionally, the illustrative fluid may further contain a non-oleaginous fluid. Such
 7 a non-oleaginous fluid preferably constitutes from about 1% to about 70% by volume of
 8 the total fluid. In one preferred and illustrative embodiment, the non-oleaginous fluid is
 9 selected from sea water, a brine containing organic or inorganic dissolved salts, a liquid
 10 containing water-miscible organic compounds, combinations of these and similar fluids
 11 used in the formulation of similar fluids. The illustrative fluid may also optionally
 12 include weighting or bridging agents. In one preferred and illustrative embodiment, the
 13 weighting or bridging agent is selected from calcium carbonate, dolomite, siderite, barite,
 14 celestite, iron oxides, manganese oxides, ulexite, carnalite, sodium chloride combinations
 15 of these and similar compounds used in the formulation of drilling and wellbore fluids.

16 In addition to the fluid compositions disclosed herein, the present invention also
 17 includes the use of such fluids in the formation, recompletion and maintenance of
 18 subterranean wells. Thus, one such illustrative embodiment includes a method of
 19 forming a subterranean well. The illustrative method includes drilling the subterranean
 20 well with a rotary drill bit and a drilling fluid that includes: an oleaginous based
 21 continuous phase and a solids tolerance agent as described above. Other methods of using
 22 the fluids disclosed are also contemplated as being part of the present invention. Such
 23 other methods and uses include using the above fluids as packing fluids, completion
 24 fluids, work-over fluids, fracturing fluids, spacer fluids and other such uses for well bore
 25 fluids that are known in the art.

26 These and other features of the present invention are more fully set forth in the
 27 following description of embodiments of the invention.
 28

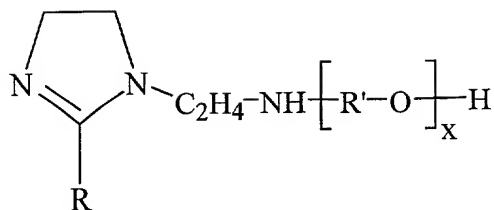
DESCRIPTION OF ILLUSTRATIVE EMBODIMENTS

The present invention is generally directed to an oil base well bore fluid that is useful in the formulation of drilling, completing and working over of subterranean wells, preferably oil and gas wells. The fluids may also be used as packing fluids, fracturing fluids and other similar well bore uses in which solids tolerance properties are desired. The fluids of the present invention may also be used to formulate invert emulsion fluids. Uses of the well bore fluids and invert emulsion fluids should be known to one of skill in the art as is noted in the book COMPOSITION AND PROPERTIES OF DRILLING AND COMPLETION FLUIDS, 5th Edition, H.C.H. Darley and George R. Gray, Gulf Publishing Company, 1988, the contents of which are hereby incorporated herein by reference.

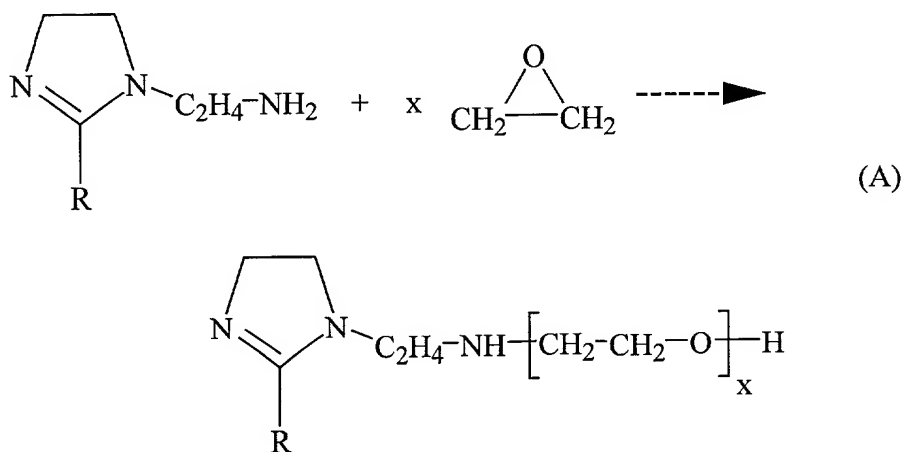
In one embodiment of the present invention, a well bore fluid is formulated to include an oleaginous fluid, and a solids tolerance agent that enables high solids loading in the fluid.

The oleaginous fluid is a liquid and more preferably is a natural or synthetic oil and more preferably the oleaginous fluid is selected from the group including diesel oil; mineral oil; a synthetic oil, such as polyolefins, polydiorganosiloxanes, siloxanes or organosiloxanes; and mixtures thereof. The concentration of the oleaginous fluid should be sufficient so that an invert emulsion forms and may be less than about 99% by volume of the invert emulsion. In one embodiment the amount of oleaginous fluid is from about 30% to about 95% by volume and more preferably about 40% to about 90% by volume of the invert emulsion fluid. The oleaginous fluid in one embodiment may include at least about 5% by volume of a material selected from the group including esters, ethers, acetals, dialkylcarbonates, hydrocarbons, and combinations thereof.

The solids tolerance agent of the present invention is believed to have the general formula:



1
2 in which R is a C₆ to C₂₀ aliphatic group and R' is a C₂ to C₆ aliphatic group and x has a
3 value from about 1 to about 10. One of ordinary skill in the art of organic synthesis by
4 applying the skills and knowledge of retro-synthesis should be able to find a variety of
5 synthetic pathways to the compounds of the present invention. However, in one preferred
6 embodiment, the solids tolerance agent is the product of the reaction of an ethylene oxide
7 with an imidazoline of a fatty acid as shown in reaction A.

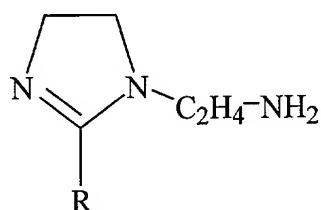


9 Reaction A can be carried out using alkylene oxides other than ethylene oxide as shown
10 in the above reaction equation. Alkylene oxide compounds are a general class of
11 compounds in which an oxo group is bound to two adjacent carbon atoms to form a
12 triangular ring. Often referred to as an epoxy group, such compounds are reactive with
13 the primary amine functionality group of the imidazoline to form the compounds useful in
14 the practice of the present invention. As previously noted the above reaction A may be
15 carried out with a wide variety of alkylene oxide compounds. In one preferred and
16 illustrative embodiment, the alkylene oxide is a C₂ to C₄ alkylene oxide and more
17 preferably the alkylene oxide is ethylene oxide or propylene oxide.

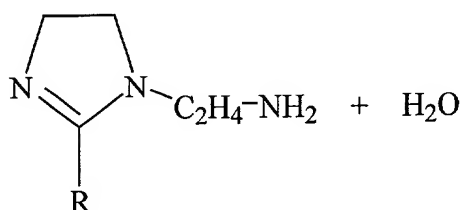
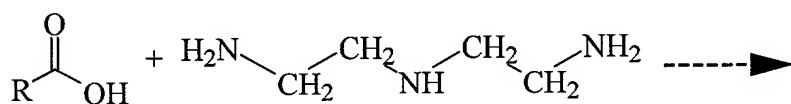
1 The above reaction A should be carried out under conditions suitable for the
2 reaction of the alkylene oxide with the imidazoline. Such conditions should be known to
3 one of ordinary skill in the art of organic chemical compound synthesis after consultation
4 with appropriate reference works.

5 In carrying out the reaction A, the molar ratio of imidazoline to alkylene oxide can
6 be widely adjusted however, preferably the molar ratio is from about 2:1 to about 1:10.
7 That is to say the value of "x" in the above reaction formula may vary from values of
8 about 0.5 to 10. Alternative synthetic pathways may be used to achieve the same target
9 compounds used in the present invention. Such alternative pathways will be apparent to
10 one of skill in the art of organic synthesis.

11 The imidazoline of a fatty acid used above in reaction A has the general formula



12 in which R is a C₆ to C₂₀ aliphatic group. We have found in one illustrative embodiment
13 that the imidazoline is preferably the reaction product of a condensation reaction of a C₆
14 to C₂₀ fatty acid and diethyltriamine as shown in Reaction B.



(B)

16 The above reaction B may be carried out with a wide variety of fatty acid compounds. In
17 one preferred and illustrative embodiment, the fatty acid is a C₁₀ to C₂₀ fatty acid and
18 more preferably the fatty acid contains at least one degree of unsaturation.

1 The above reaction B should be carried out under conditions suitable for the
2 condensation reaction of the diethyltriamine with fatty acid. Such conditions should be
3 known to one of ordinary skill in the art of organic chemical compound synthesis after
4 consultation with appropriate reference works.

5 In carrying out the reaction B, the molar ratio of diethyltriamine to fatty acid can
6 be widely adjusted however, preferably the molar ratio is from about 1:1. As with the
7 final compounds, a skilled person in the art of organic synthesis should be able to reveal a
8 variety of synthetic pathways of creating the desired imidazoline.

9 The amount of solids tolerance agent present in the drilling fluids of the present
10 invention, as noted above, should be sufficient to increase the solids loading in the
11 drilling fluid. While the concentration of the solids tolerance agent may vary depending
12 on the particular components in the drilling fluid or mud, typically the concentration is
13 less than about 10 % by volume of the fluid. Thus in one embodiment the solids
14 tolerance agent is preferably present in the drilling fluid at a concentration of about 0.1%
15 to about 5.0%. More preferably the amount of solids tolerance agent in the drilling fluid
16 should be present in a concentration of about 1 to about 5% by volume of the fluid.

17 The above combination of oleaginous fluid and solids tolerance agent may be
18 used to formulate invert emulsions with non-oleaginous fluids. In one such embodiment,
19 the non-oleaginous fluid used in the formulation of the invert emulsion fluid is a liquid
20 and preferably is an aqueous liquid. More preferably, the non-oleaginous liquid may be
21 selected from the group including sea water, a brine containing organic and/or inorganic
22 dissolved salts, liquids containing water-miscible organic compounds, combinations of
23 these and similar compounds used in the formulation of invert emulsions. The amount of
24 the non-oleaginous fluid is typically less than the theoretical limit needed for forming an
25 invert emulsion. Thus in one embodiment the amount of non-oleaginous fluid is less than
26 about 70% by volume and preferably from about 1% to about 70% by volume. In another
27 embodiment, the non-oleaginous fluid is preferably from about 5% to about 60% by
28 volume of the invert emulsion fluid.

29 In yet another embodiment of the present invention, other surfactant compounds
30 may be used in conjunction with the solids tolerance agents utilized herein. In such cases

1 it is important however that the quantity and nature of these supplemental surfactants
2 should not interfere in the ability and properties given the emulsion fluid by the solids
3 tolerance agent to act as described herein.

4 The fluids of the present invention are especially useful in the drilling,
5 completion and working over of subterranean oil and gas wells. In particular the fluids
6 of the present invention are useful in formulating drilling muds and completion fluids
7 that require a high loading of solids content in the drilling fluid. Such muds and fluids
8 are especially useful in the drilling of horizontal wells into hydrocarbon bearing
9 formations.

10 The method used in preparing the drilling fluids of the present invention is not
11 critical. Conventional methods can be used to prepare the drilling fluids of the present
12 invention in a manner analogous to those normally used, to prepare conventional oil-
13 based drilling fluids. In one representative procedure, a desired quantity of oleaginous
14 fluid such as a base oil and a suitable amount of the solids tolerance agent are mixed
15 together and the remaining components are added sequentially with continuous mixing.
16 An invert emulsion based on this fluid may be formed by vigorously agitating, mixing or
17 shearing the oleaginous fluid with a non-oleaginous fluid.

18 The fluids of the present invention may further contain additional components
19 depending upon the end use of the invert emulsion so long as they do not interfere with
20 the functionality of the solids tolerance agents described herein. For example, wetting
21 agents, organophilic clays, viscosifiers, weighting agents, bridging agents and fluid loss
22 control agents may be added to the fluid compositions of this invention for additional
23 functional properties. The addition of such agents should be well known to one of skill in
24 the art of formulating drilling fluids and muds.

25 Wetting agents that may be suitable for use in this invention include, crude tall
26 oil, oxidized crude tall oil, surfactants, organic phosphate esters, modified imidazolines
27 and amidoamines, alkyl aromatic sulfates and sulfonates, and the like, and
28 combinations or derivatives of these. However, the use of fatty acid wetting agents
29 should be minimized so as to not adversely effect the reversibility of the invert
30 emulsion disclosed herein. Versawet® and Versawet®NS are examples of

1 commercially available wetting agents manufactured and distributed by M-I L.L.C. that
2 may be used in this invention. Silwet L-77, L-7001, L7605 and L-7622 are examples
3 of commercially available surfactants and wetting agents manufactured and distributed
4 by Union Carbide Chemical Company Inc.

5 Organophilic clays, normally amine treated clays, may be useful as viscosifiers
6 in the fluid compositions of the present invention. Other viscosifiers, such as oil
7 soluble polymers, polyamide resins, polycarboxylic acids and soaps can also be used.
8 The amount of viscosifier used in the composition can vary depending upon the end use
9 of the composition. However, normally about 0.1% to 6% by weight range are
10 sufficient for most applications. VG-69 and VG-PLUS are organoclay materials
11 distributed by M-I L.L.C., and Versa-HRP is a polyamide resin material manufactured
12 and distributed by M-I L.L.C., that may be used in this invention.

13 Conventional suspending agents that may be utilized in this invention include
14 organophilic clays, amine treated clays, oil soluble polymers, polyamide resins,
15 polycarboxylic acids, and soaps. The amount of conventional suspending agent used in
16 the composition, if any, may vary depending upon the end use of the composition.
17 However, normally about 0.1 % to about 6 % by weight is sufficient for most
18 applications. VG-69 and VG-PLUS are organoclay materials distributed by M-I
19 L.L.C., and Versa-HRP is a polyamide resin material manufactured and distributed by
20 M-I L.L.C., that may be used in this invention.

21 Weighting agents or density materials suitable for use in this invention include
22 galena, hematite, magnetite, iron oxides, illmenite, barite, siderite, celestite, dolomite,
23 calcite, and the like. The quantity of such material added, if any, depends upon the
24 desired density of the final composition. Typically, weight material is added to result
25 in a drilling fluid density of up to about 24 pounds per gallon. The weight material is
26 preferably added up to 21 pounds per gallon and most preferably up to 19.5 pounds per
27 gallon.

28 Fluid loss control agents typically act by coating the walls of the borehole as the
29 well is being drilled. Suitable fluid loss control agents which may find utility in this
30 invention include modified lignites, asphaltic compounds, gilsonite, organophilic

1 humates prepared by reacting humic acid with amides or polyalkylene polyamines, and
 2 other non-toxic fluid loss additives. Typically, fluid loss control agents are added in
 3 amounts less than about 10% and preferably less than about 5% by weight of the fluid.

4 The following examples are included to demonstrate preferred embodiments of
 5 the invention. It should be appreciated by those of skill in the art that the techniques and
 6 compositions disclosed in the examples which follow represent techniques discovered by
 7 the inventors to function well in the practice of the invention, and thus can be considered
 8 to constitute preferred modes for its practice. However, those of skill in the art should, in
 9 light of the present disclosure, appreciate that many changes can be made in the specific
 10 embodiments which are disclosed and still obtain a like or similar result without
 11 departing from the scope of the invention.

12 General Information Relevant to the Examples

13 These tests were conducted in accordance with the procedures in API Bulletin RP
 14 13B-2, 1990. The following abbreviations are sometimes used in describing the results of
 15 experimentation.

16 "PV" is plastic viscosity which is one variable used in the calculation of viscosity
 17 characteristics of a drilling fluid, measured in centipoise (cp) units.

18 "YP" is yield point which is another variable used in the calculation of viscosity
 19 characteristics of drilling fluids, measured in pounds per 100 square feet (lb/100 ft²).

20 "AV" is apparent viscosity which is another variable used in the calculation of
 21 viscosity characteristic of drilling fluid, measured in centipoise (cp) units.

22 "GELS" is a measure of the suspending characteristics, or the thixotropic
 23 properties of a drilling fluid, measured in pounds per 100 square feet (lb/100 ft²).

24 "API F.L." is the term used for API filtrate loss in milliliters (ml).

25 "HTHP" is the term used for high temperature high pressure fluid loss, measured
 26 in milliliters (ml) according to API bulletin RP 13 B-2, 1990.

27 Example 1.

28 A base mud was prepared having the following formulation:
 29

Component	Amount (g)
C ₁₆ -C ₁₈ Internal olefin	160.4
25% wt Calcium Chloride brine	90.5
Nova Mul L	6
Nova Wet	2
Lime	5
VG Plus	4
Barite	25

1
2 A measured amount of the compounds of the present invention (CP1 and CP2)
3 and Rev Dust (pounds per barrel, ppb), a drill cuttings solids substitute, were added to the
4 base fluid and the mixture was agitated. CP1 is the reaction product of 1 mole of
5 propylene oxide with 1 mole of the imidazoline of a C₁₆-C₁₈ fatty acid. CP2 is the
6 reaction product of 2 moles of propylene oxide with 1 mole of the imidazoline of C₁₆-C₁₈
7 fatty acid. For comparison purposes a control fluid was formulated without the addition
8 of the compounds of the present invention. The mixtures were then heat aged in rotating
9 ovens at 250 F for 16 hours. The properties of the resulting fluids were measure and
10 representative data is given in Table A below.

1

TABLE A

Rev Dust (ppb)		25	100	150	175
CP1	PV	23	52	60	81
	YP	14	31	34	9
	Gels, 10 Min.	12	17	17	10
	ES	270	182	151	83
CP1	PV	25	50	56	88
	YP	11	24	24	37
	Gels, 10 Min.	12	12	12	18
	ES	300	191	144	103
Control	PV	27	56	33	--
	YP	13	31	34	--
	Gels, 10 Min.	13	17	18	8
	ES	257	178	146	43

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Upon review of the above representative data, one of ordinary skill in the art should appreciate that the addition of CP1 and CP2 enable higher solids loading of the aged drilling fluid as compared to the control fluid.

Example 2.

A base mud with 75:25 oil:water ratio using a 25% calcium chloride brine was prepared having the following formulation:

Component	Amount (g)
C ₁₆ -C ₁₈ Internal olefin	160.4
25% wt Calcium Chloride brine	90.5
Nova Mul L	6.0
Nova Wet	2.0
Lime	5.0
VG Plus	4.0
Barite	211.4

1
 2 A measured amount of the compounds of the present invention having a variable
 3 extent of reaction with alkylene oxide was added and Rev Dust (pounds per barrel, ppb),
 4 a drill cuttings solids substitute, were added to the base fluid and the mixture was
 5 agitated. The compounds GS 1956A, Gs 1956B, GS1957A, GS1957B, GS1809A,
 6 GS1809B, GS1810A, GS1810B, GS1959, GS1960 are commercially available from
 7 Champion Chemicals. The following table provides
 8

Compound	Equivalents of alkylene oxide reacted with imidazoline
Control	0.0
CP3 (0.5 EO-GS1956A)	0.5 ethylene oxide
CP4 (1.0 EO-GS1956B)	1.0 ethylene oxide
CP5 (1.5 EO – GS 1957A)	1.5 ethylene oxide
CP6 (2.0 EO – GS1957B)	2.0 ethylene oxide
CP7 (0.38 PO - GS1809A)	0.38 propylene oxide
CP8 (0.75 PO – GS1809B)	0.75 propylene oxide
CP9 (1.125 PO – GS1810A)	1.125 propylene oxide
CP10 (1.5 PO – GS1810B)	1.5 propylene oxide
CP11 (2.0 PO – GS 1810B)	2.0 propylene oxide
CP12 (5.0 PO – GS 1959)	5.0 propylene oxide
CP13 (10.0 PO – GS 1960)	10.0 propylene oxide

1
2 The mixtures were then heat aged in rotating ovens at 150 F for 16 hours. The
3 properties of the resulting fluids were measure at 120 F and representative data is given in
4 Tables B, C & D below.

TABLE B: 25 g Rev Dust – Heat Aged

Fluid Properties (at 120°F)	Control	CP3	CP4	CP5	CP6	CP7	CP8	CP9	CP10	CP11	CP12	CP13
Rheology												
600	52	55	49	56	55	60	62	61	61	53	57	61
300	30	31	28	32	31	37	37	35	36	30	32	35
200	23	24	22	25	24	27	28	27	27	24	26	28
100	15	15	14	16	15	19	19	17	18	15	16	17
6	5	5	5	5	5	7	7	5	7	5	5	5
3	4	4	4	4	4	5	5	4	6	4	4	4
PV	22	24	21	24	24	23	25	26	25	23	25	26
YP	8	7	7	8	7	14	12	9	11	7	7	9
Gels												
10 sec.	5	6	5	5	6	7	7	6	7	6	6	6
10 min.	6	6	5	5	7	12	12	11	12	7	7	8
ES	261	229	238	284	217	270	261	220	300	251	217	221

TABLE C: 125 g Rev Dust – Heat Aged

Fluid Properties (at 120°F)	Control	CP3	CP4	CP5	CP6	CP7	CP8	CP9	CP10	CP11	CP12	CP13
Rheology												
600	96	87	61	62	61	147	140	161	136	61	260	--
300	57	52	35	37	36	90	85	95	81	35	147	175
200	42	38	27	28	27	69	65	71	61	28	115	170
100	26	22	17	19	18	45	41	46	38	17	68	130
6	7	6	5	7	7	15	13	13	11	5	18	28
3	6	5	4	5	6	13	11	11	9	4	16	25
PV	39	37	26	25	25	57	55	66	55	26	113	--
YP	18	15	9	12	11	33	30	29	26	9	34	--
Gels												
10 sec.	7	7	6	7	7	15	13	13	10	6	28	100
10 min.	9	8	11	12	12	18	15	15	12	9	31	110
ES	143	116	220	261	300	165	173	165	188	274	142	48

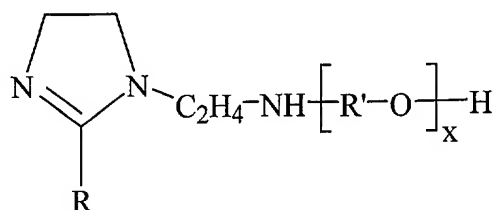
TABLE D: 200 g Rev Dust – Heat Aged

Fluid Properties (at 120°F)	Control	CP3	CP4	CP5	CP6	CP11	CP12	CP13
Rheology								
600	OS	OS	OS	OS	OS	OS	OS	OS
300	190	127	112	101	87	86	OS	OS
200	30	27	27	28	27	26	71	79
100	20	19	17	19	18	18	20	24
6	6	7	5	7	7	7	20	23
3	5	5	4	5	6	7	--	21
Gels								
10 sec.	7	7	6	7	7	7	20	28
10 min.	13	12	11	12	12	11	20	30
ES	119	111	156	165	187	200	65	21

Note: "OS" indicates and off-scale reading

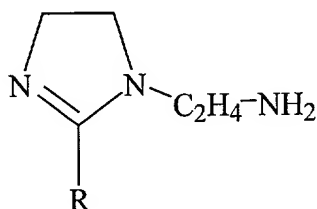
1 celestite, iron oxides, manganese oxides, ulexite, carnalite, sodium chloride combinations
2 of these and similar compounds used in the formulation of drilling and wellbore fluids.

3 Another illustrative embodiment of the present invention that should be
4 appreciated by one of skill in the art is an invert emulsion fluid having utility for drilling,
5 completing, or working over subterranean wells. Such an illustrative fluid is formulated
6 to include: an oleaginous liquid; a non-oleaginous liquid; and a solids tolerance agent
7 having the formula:



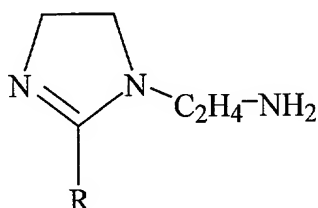
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9 in which R is a C₆ to C₂₀ aliphatic group and R' is a C₂ to C₆ aliphatic group and x has a
10 value from about 1 to about 10. The invert emulsion is formulated such that the
11 oleaginous liquid is from about 30% to about 99% by volume of said fluid, the non-
12 oleaginous liquid is from about 1% to about 70% by volume of said fluid, and the solids
13 tolerance agent is present in said fluid at a concentration of about 0.1% to about 5.0% by
14 weight of the total fluid. The solids tolerance agent is preferably formulated such that the
15 R' group is ethyl or isopropyl. Alternatively, the solids tolerance agent can be formulated
16 such that the R group is unsaturated. That is to say the R group has at least one carbon-
17 carbon double bond. In one preferred and illustrative embodiment the oleaginous fluid is
18 formulated such that from about 5 to about 100% by volume of the oleaginous fluid is a
19 material selected from diesel oil, mineral oil, a synthetic oil, esters, ethers, acetals, di-
20 alkylcarbonates, hydrocarbons, combinations of these and similar fluids used in the
21 formulation of invert emulsion drilling fluids. The non-oleaginous liquid in one
22 illustrative embodiment is preferably composed of sea water, a brine containing organic
23 or inorganic dissolved salts, a liquid containing water-miscible organic compounds,
24 combinations of these and similar fluids used in the formulation of invert emulsion
25 drilling fluids.

1 A third illustrative embodiment of the present invention is a fluid having utility in
2 subterranean wells. Such an illustrative fluid is formulated to include an oleaginous fluid
3 and a solids tolerance agent. The solids tolerance agent is the product of the reaction of
4 an alkylene oxide with an imidazoline of a fatty acid having the formula



5
6 in which R is a C₆ to C₂₀ aliphatic group. The imidazoline is preferably the reaction
7 product of a condensation reaction of a C₆ to C₂₀ fatty acid and diethyltriamine. In one
8 preferred and illustrative embodiment, the alkylene oxide is a C₂ to C₄ alkylene oxide and
9 more preferably the alkylene oxide is ethylene oxide or propylene oxide. In carrying out
10 the reaction, the molar ratio of imidazoline to alkylene oxide can be from about 2:1 to
11 about 1:10. In one preferred and illustrative embodiment, the oleaginous fluid is from
12 about 30% to about 99% by volume of the total fluid, and preferably is composed from
13 about 5% to about 100% by volume of a material selected from diesel oil, mineral oil,
14 synthetic oil, esters, ethers, acetals, di-alkylcarbonates, olefins, combinations of these and
15 similar fluids used in the formulation of drilling fluids. For another alternative and
16 illustrative embodiment, the fluid may be formulated so as to include a non-oleaginous
17 fluid that is preferably from about 1% to about 70% by volume of the total fluid. When a
18 non-oleaginous fluid is included, it preferably is selected from sea water, a brine
19 containing organic or inorganic dissolved salts, a liquid containing water-miscible organic
20 compounds, combinations of these and similar fluids used in the formulation of drilling
21 fluids. Optionally the illustrative fluid may include a weighting agent or a bridging agent.
22 Preferably the weighting or bridging agent is selected from calcium carbonate, dolomite,
23 siderite, barite, celestite, iron oxides, manganese oxides, ulexite, carnalite, sodium
24 chloride, and mixtures of these and similar agents used in the formulation of drilling
25 fluids.

1 In addition to the fluid compositions disclosed above, one of skill in the art should
2 understand and appreciate that the present invention also includes the use of such fluids.
3 Thus, one such illustrative embodiment includes a method of forming a subterranean
4 well. The illustrative method includes drilling the subterranean well with a rotary drill bit
5 and a drilling fluid that includes: an oleaginous based continuous phase and a solids
6 tolerance agent as described above. It is preferred that the solids tolerance agent is the
7 product of the reaction of an alkylene oxide with an imidazoline of a fatty acid having the
8 formula



9
10 in which R is a C₆ to C₂₀ aliphatic group. It is further preferred that the imidazoline is the
11 reaction product of a condensation reaction of a C₆ to C₂₀ fatty acid and diethyltriamine.
12 In such an illustrative embodiment, the alkylene oxide is preferably a C₂ to C₄ alkylene
13 oxides and more preferably ethylene oxide or propylene oxide. In carrying out the
14 reaction between the amine and the alkylene oxide, the molar ratio of imidazoline to
15 alkylene oxide is preferably from about 2:1 to about 1:10.

16 Other methods of using the fluids disclosed are also contemplated as being part of
17 the present invention. Such alternative methods and uses include using the above fluids
18 as packing fluids, completion fluids, work-over fluids, fracturing fluids, spacer fluids and
19 other such uses for well bore fluids that are known in the art.

20 While the apparatus, compositions and methods of this invention have been
21 described in terms of preferred or illustrative embodiments, it will be apparent to those of
22 skill in the art that variations may be applied to the process described herein without
23 departing from the concept and scope of the invention. All such similar substitutes and
24 modifications apparent to those skilled in the art are deemed to be within the scope and
25 concept of the invention as it is set out in the following claims.